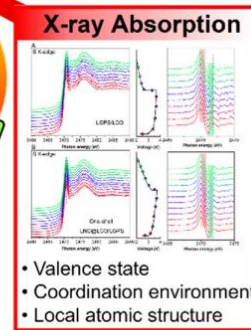
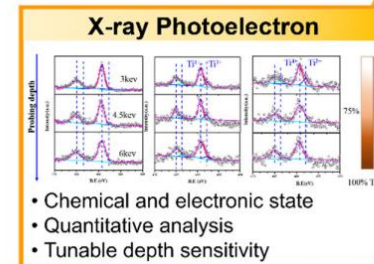
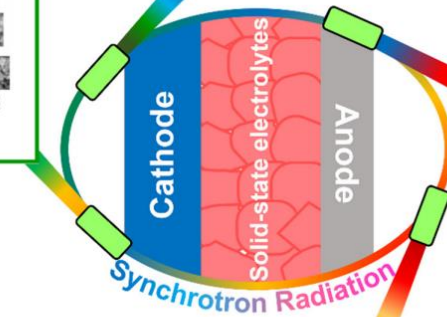
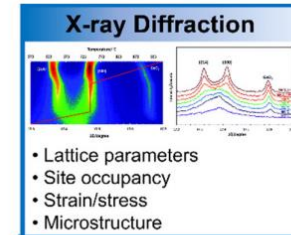
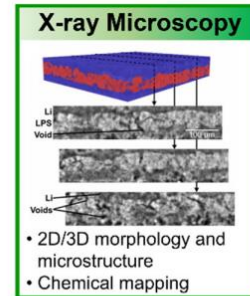
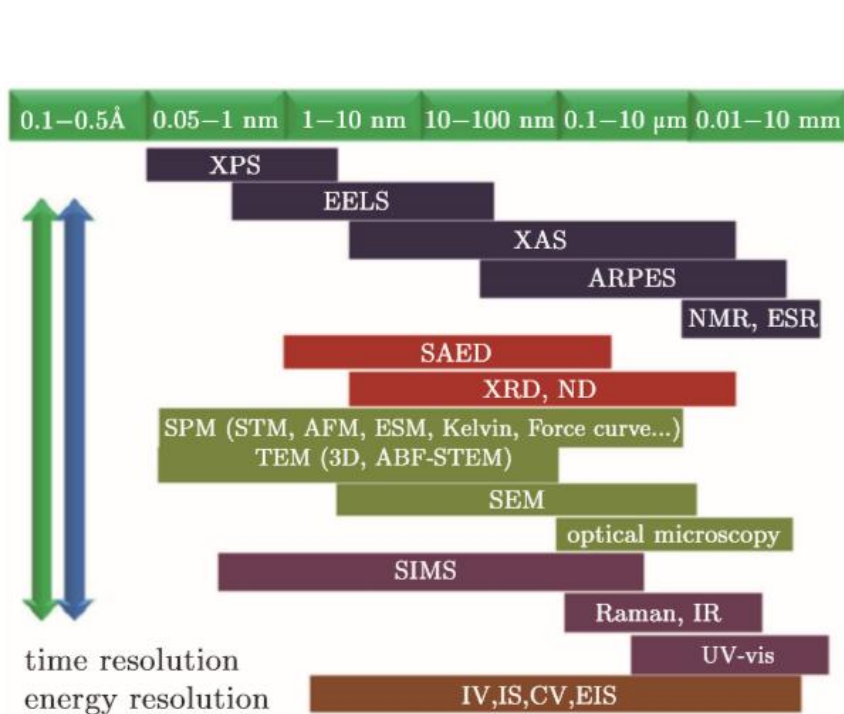


Operando powder X-ray diffraction *in application to battery materials*

STANISLAV S. FEDOTOV / IVAN A. TRUSSOV

Characterization techniques



Terminology

Ex situ

1. Run electrochemical experiment



2. Stop experiment



3. Cell disassembling



4. Analysis

In a conventional cell

In situ

1. Run electrochemical experiment



2. Stop experiment



3. Analysis

In an special electrochemical cell

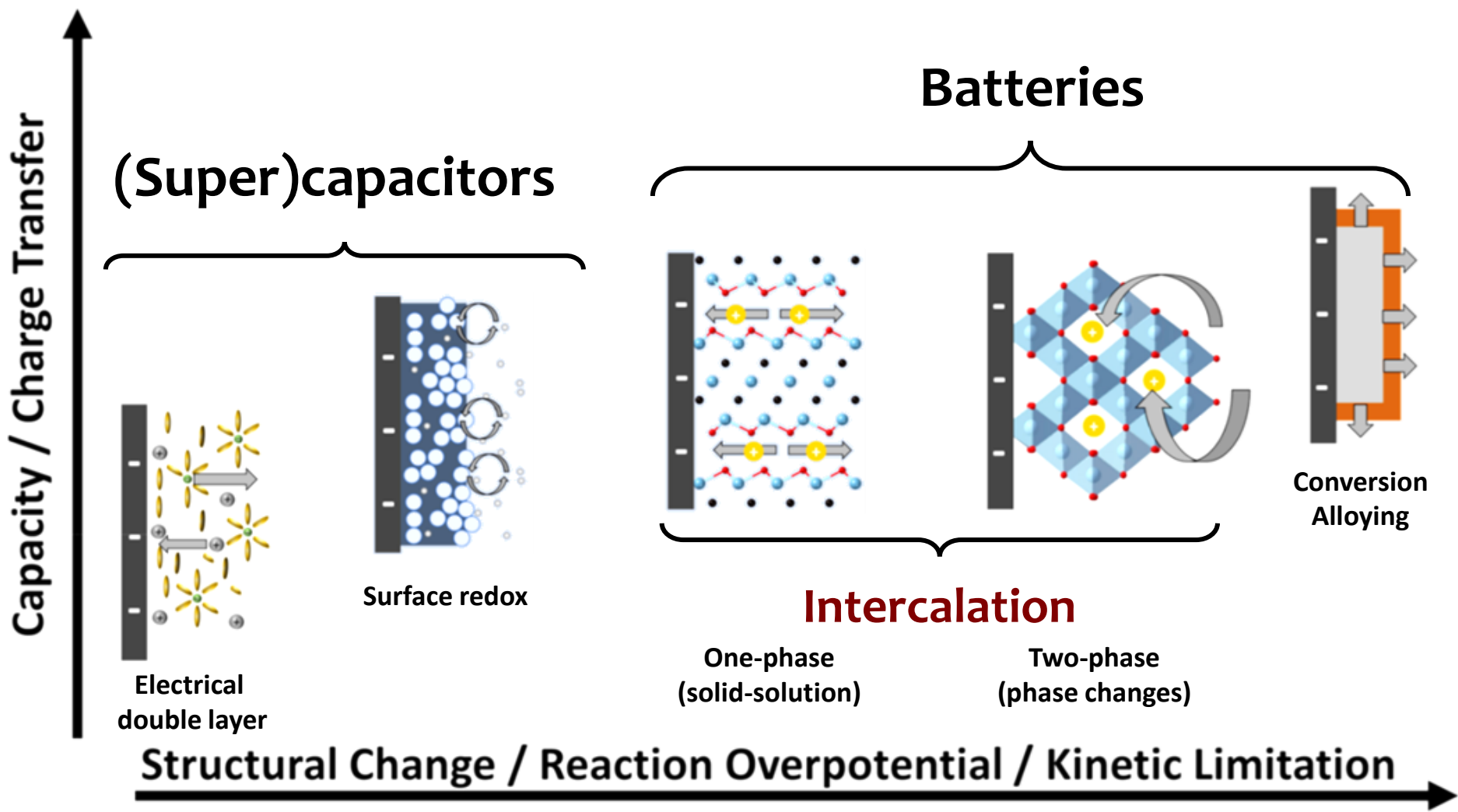
Operando

1. Run electrochemical experiment



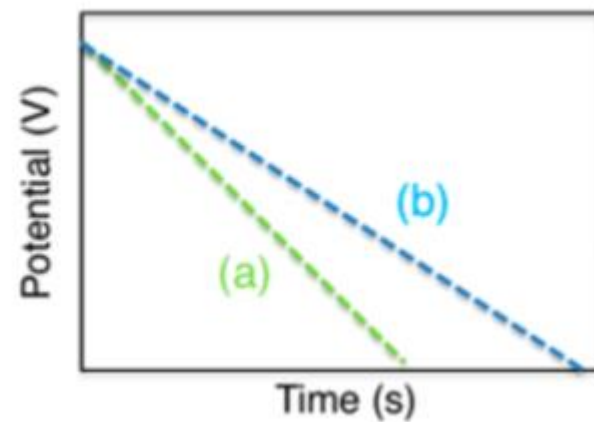
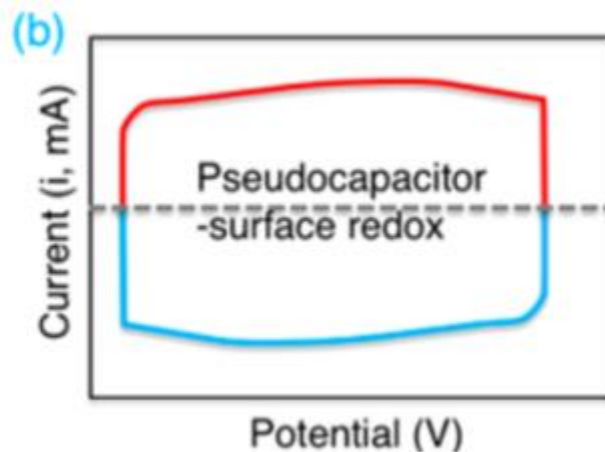
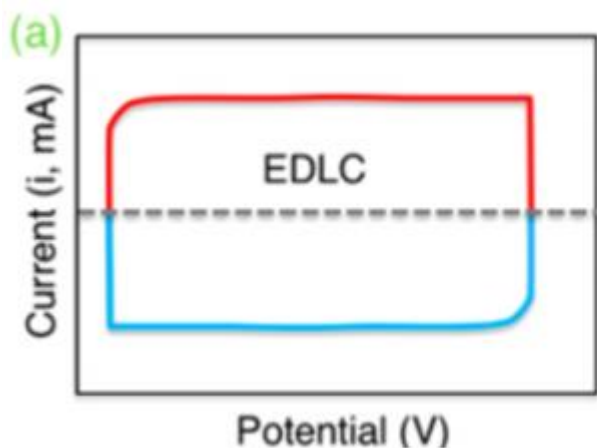
1. Analysis

Electrochemical energy storage mechanisms

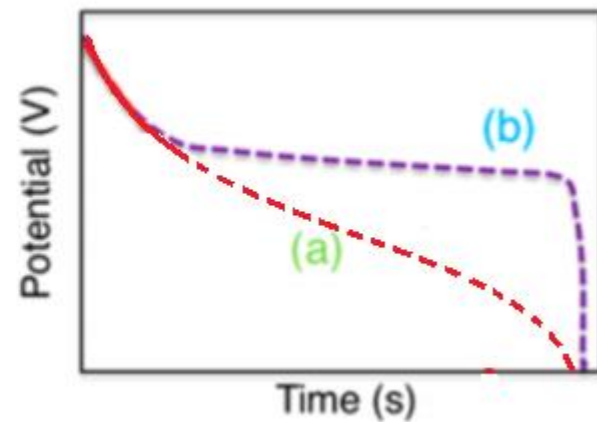
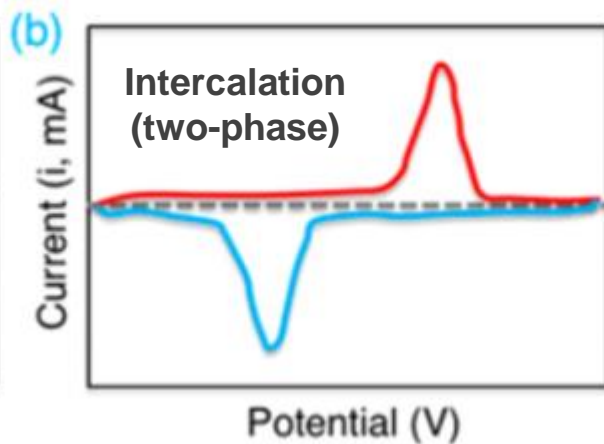
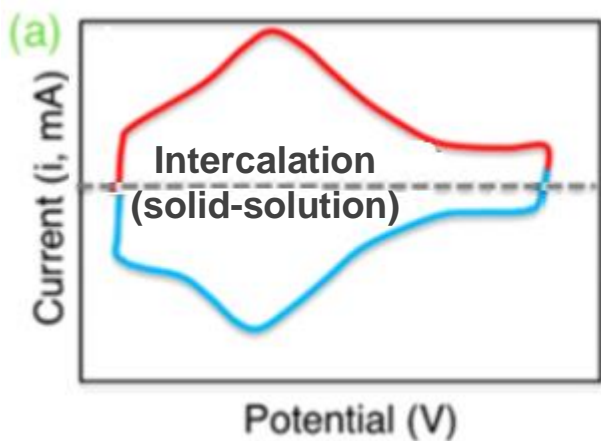


Electrochemical energy storage mechanisms

(Super)capacitors

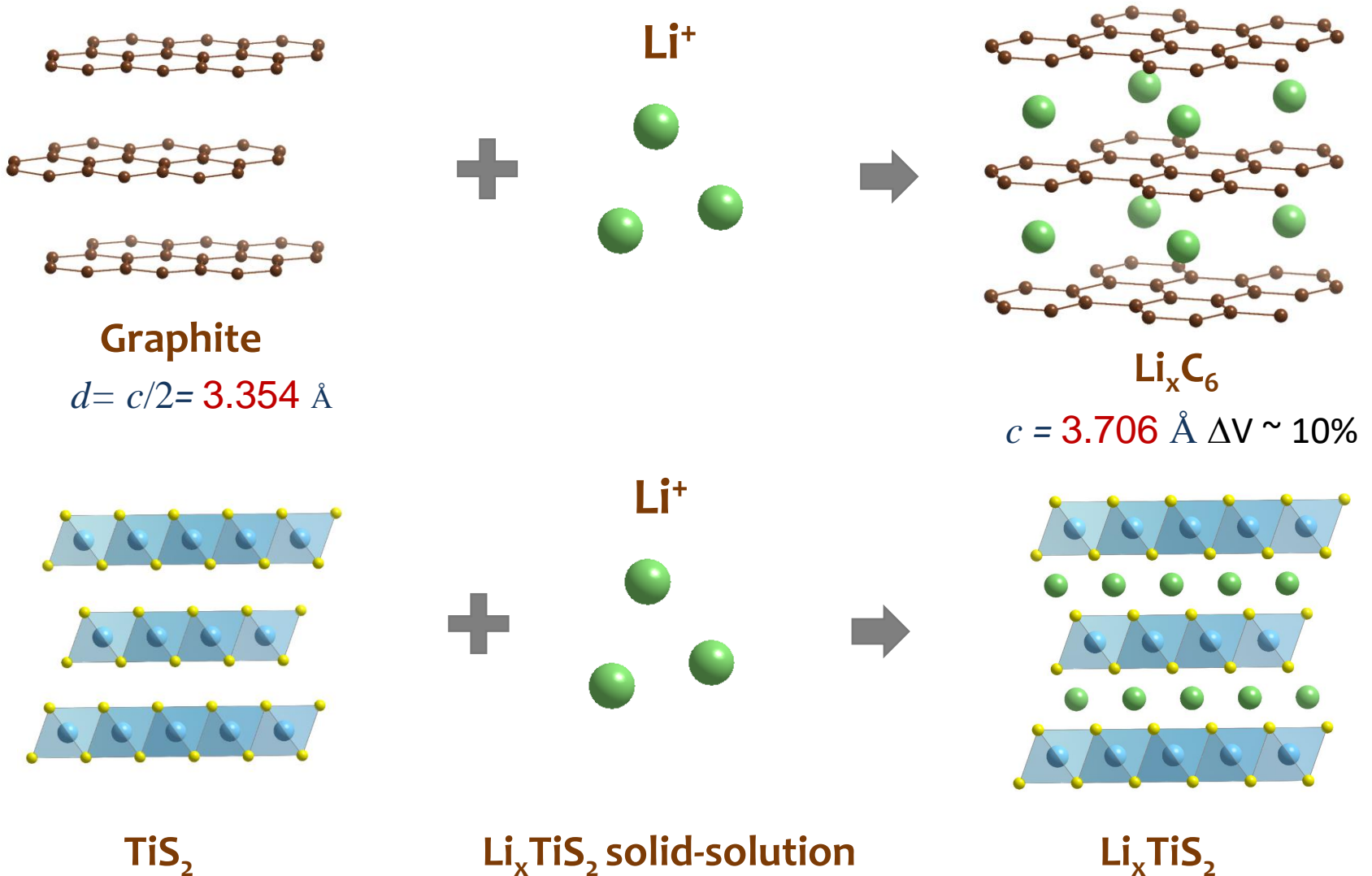


Batteries

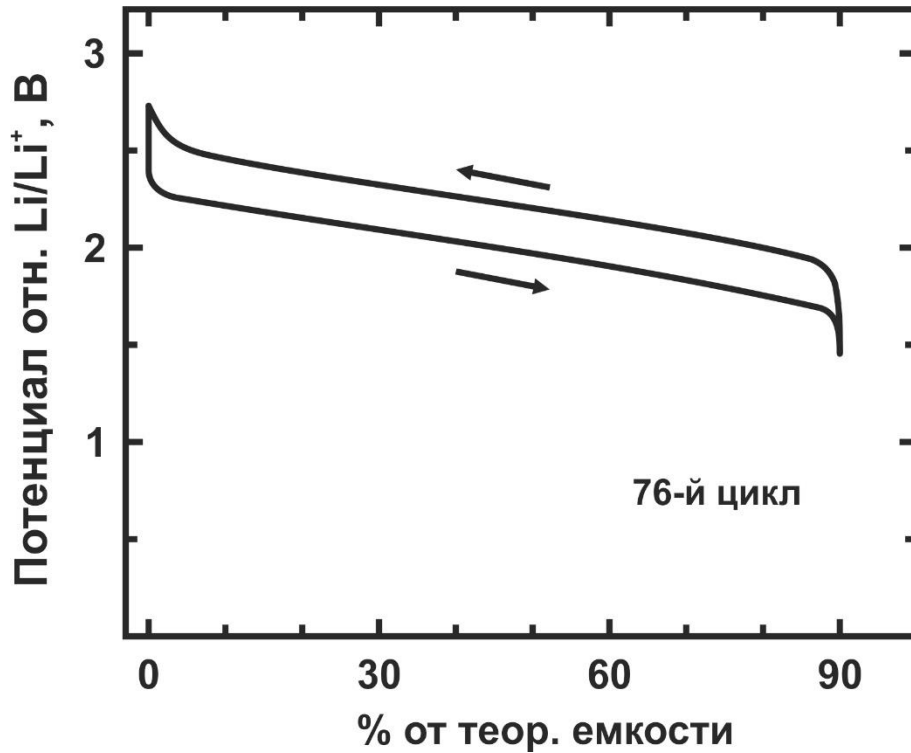


Graphite intercalation compounds (GICs)

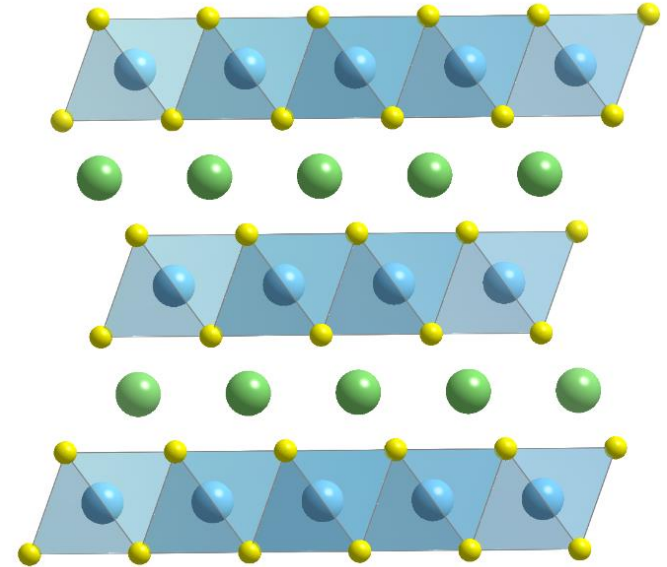
Intercalation, 1841 – graphite bisulfate



Intercalation cathode materials

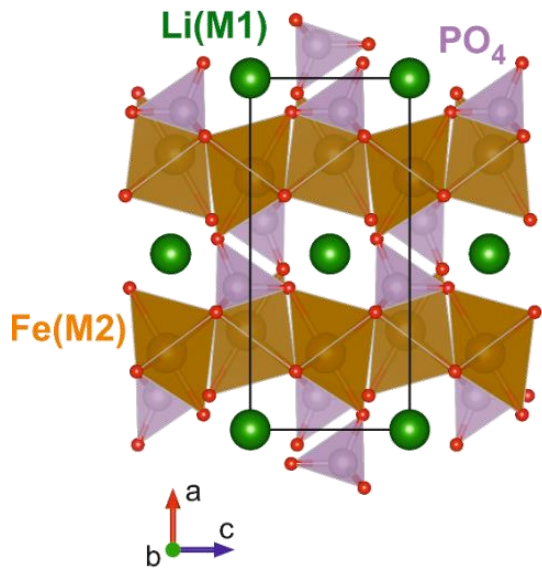


Potential is around 2.0 V vs Li/Li⁺
Li/TiS₂ battery by EXXON in 1980-s

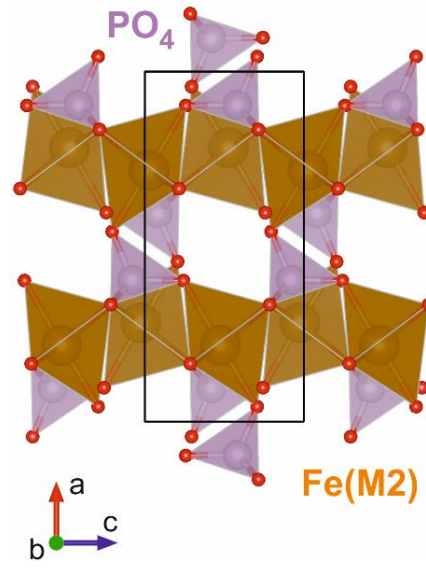


Li_xTiS_2 solid-solution

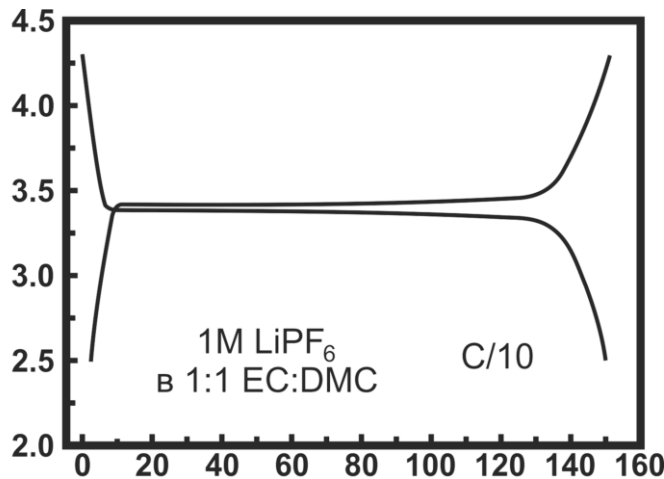
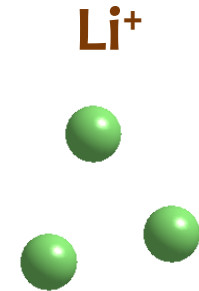
Intercalation cathode materials



LiFePO_4 Triphylite



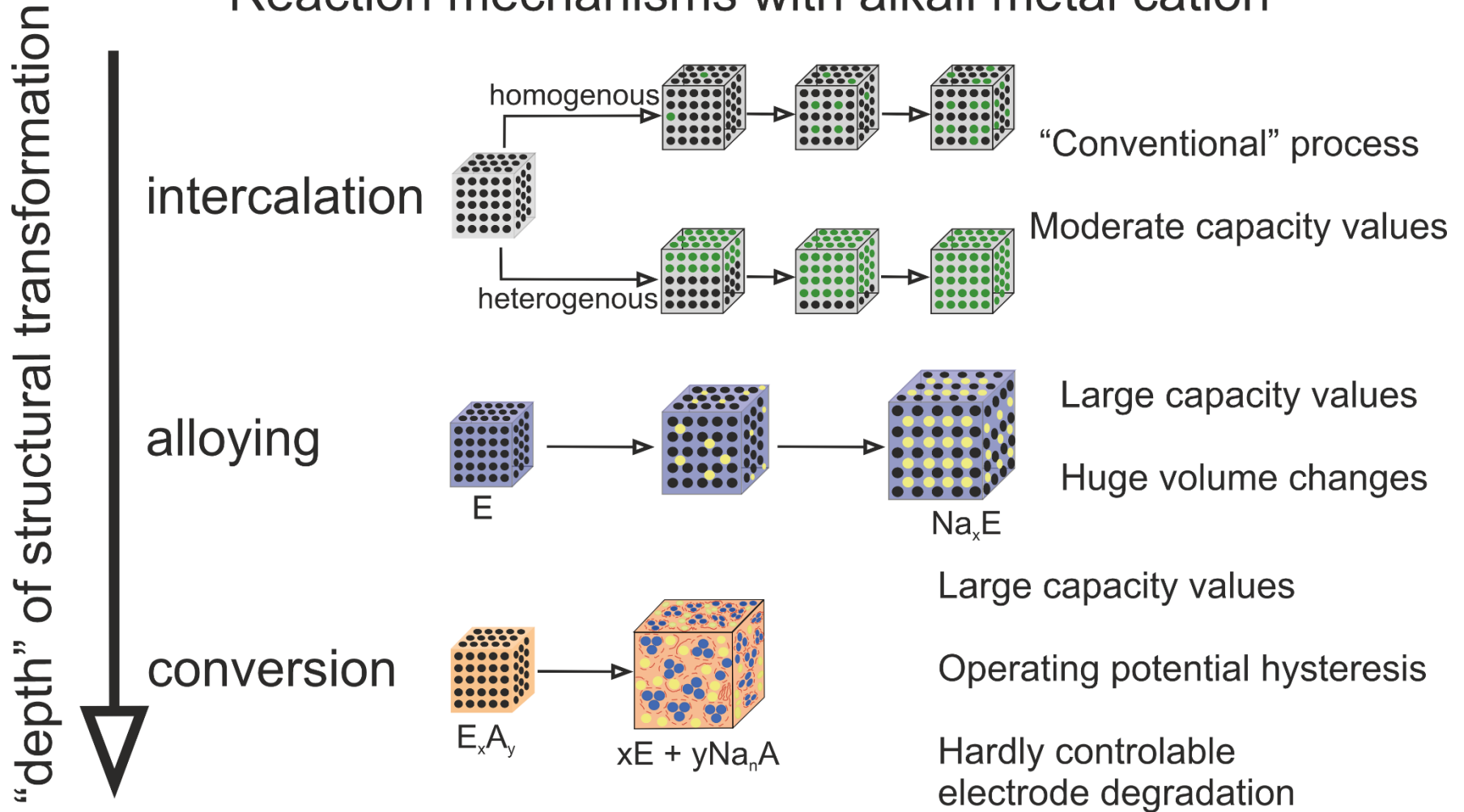
FePO_4 Heterosite



$\sim 3.43 \text{ V vs. Li}^+/\text{Li}$
Two-phase mechanism

Electrochemical energy storage mechanisms

Reaction mechanisms with alkali metal cation

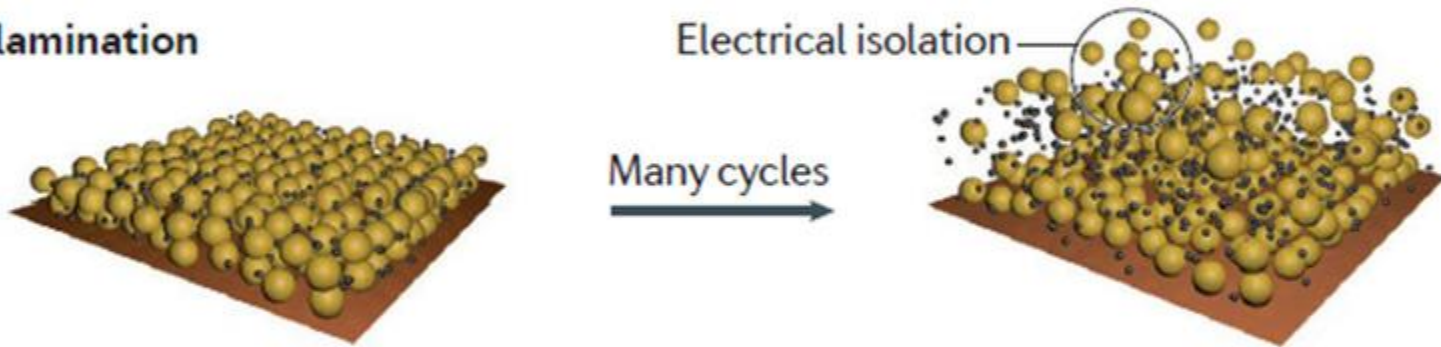


Conversion and alloying

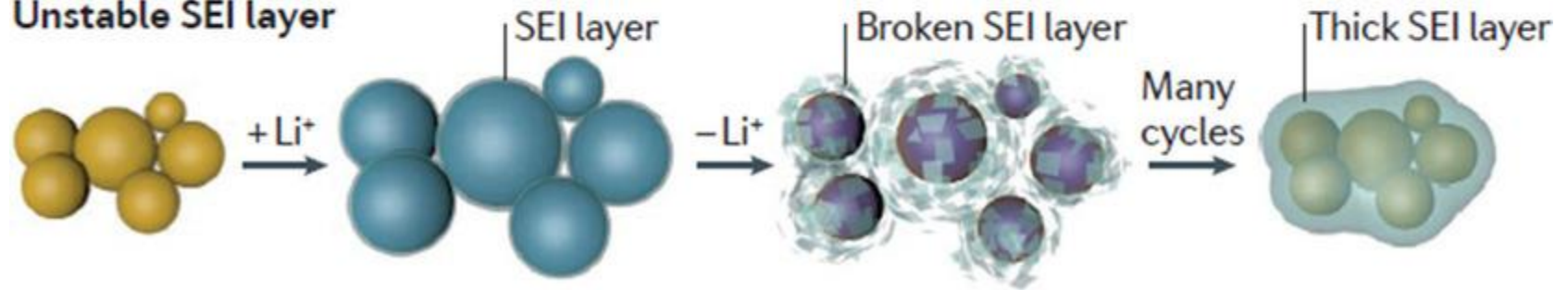
Pulverization



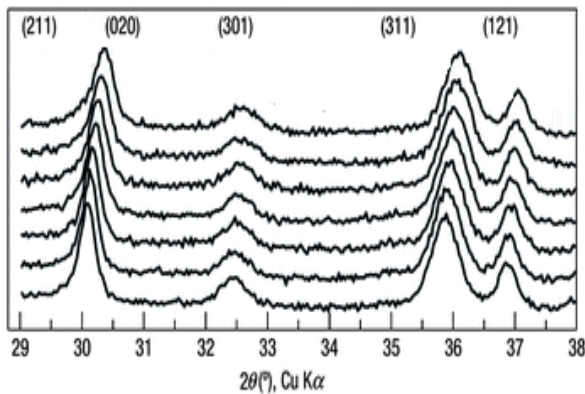
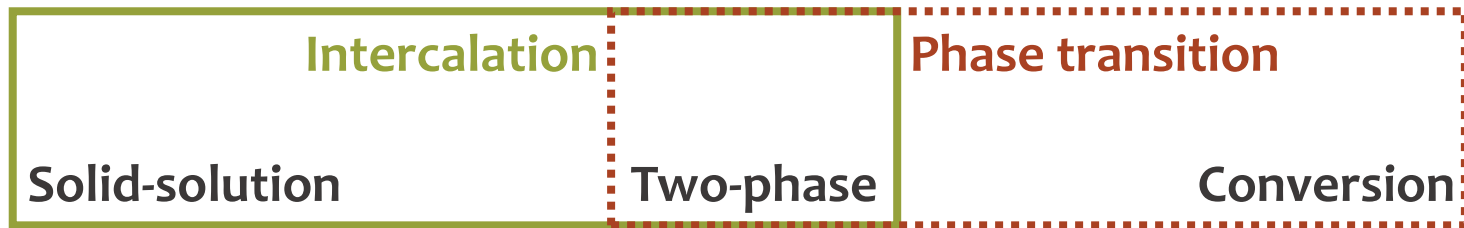
Delamination



Unstable SEI layer



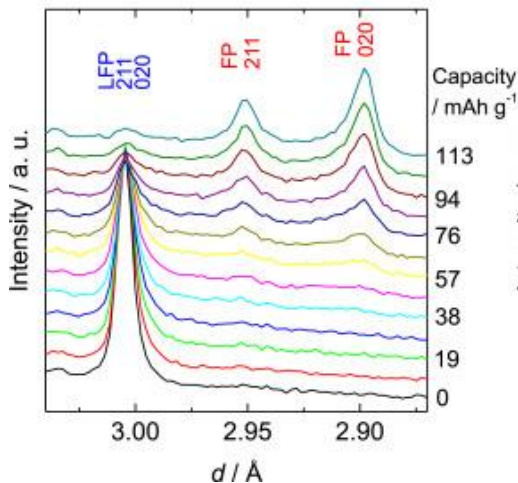
XRD changes vs. de/intercalation mechanism



Gradual change of cell parameters

Symmetry preserves

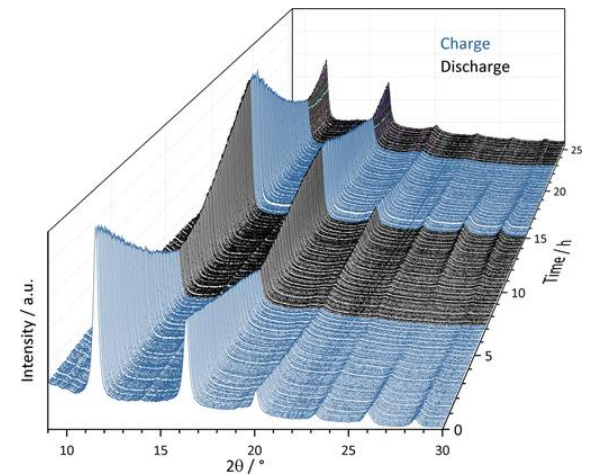
Rapid kinetics



Two phases co-exist with different cell parameters

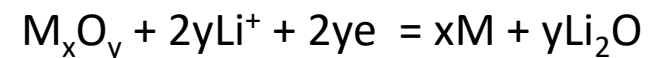
Symmetry may change

Phase boundary propagation

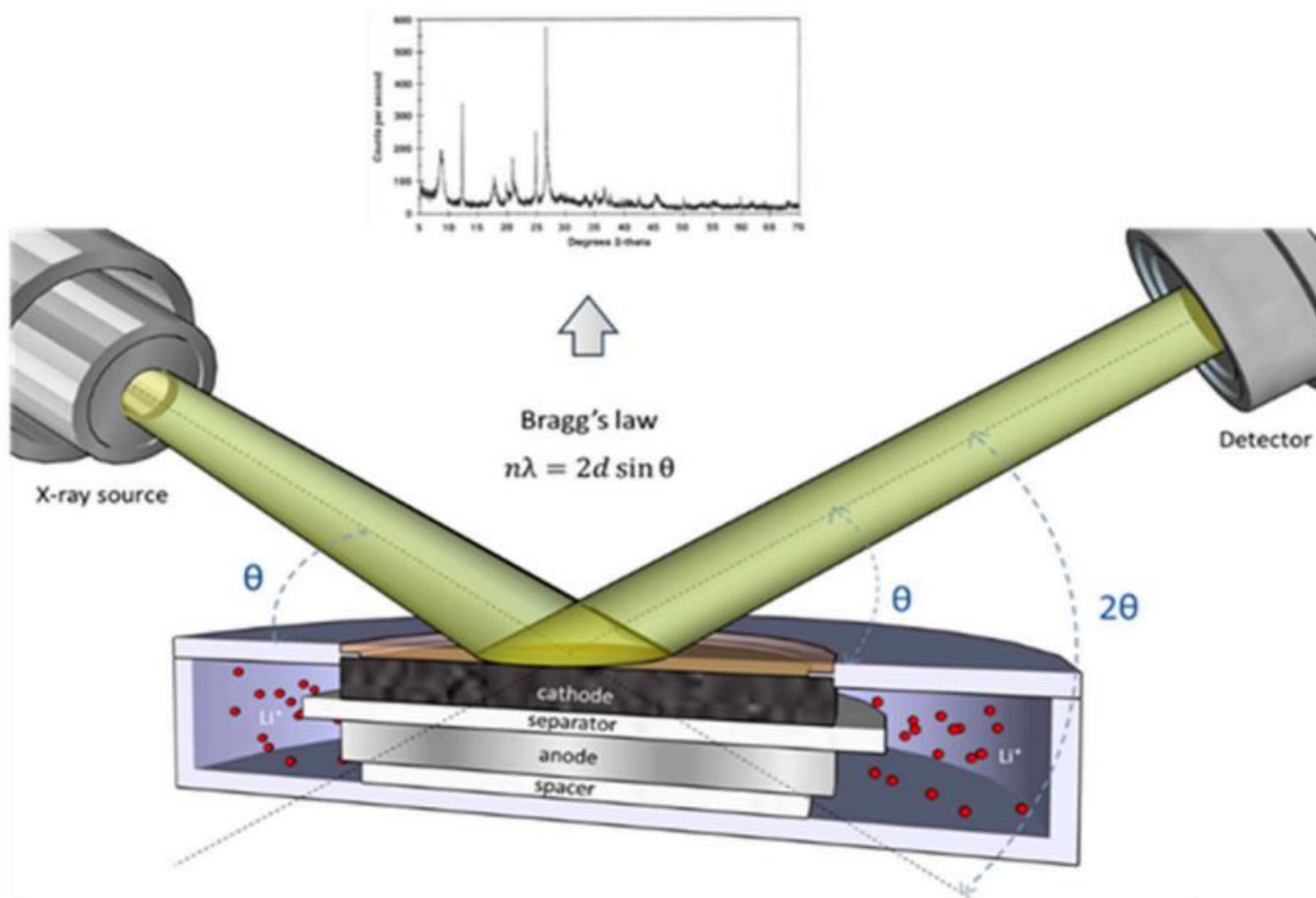


Radically different crystal structure and cell parameters

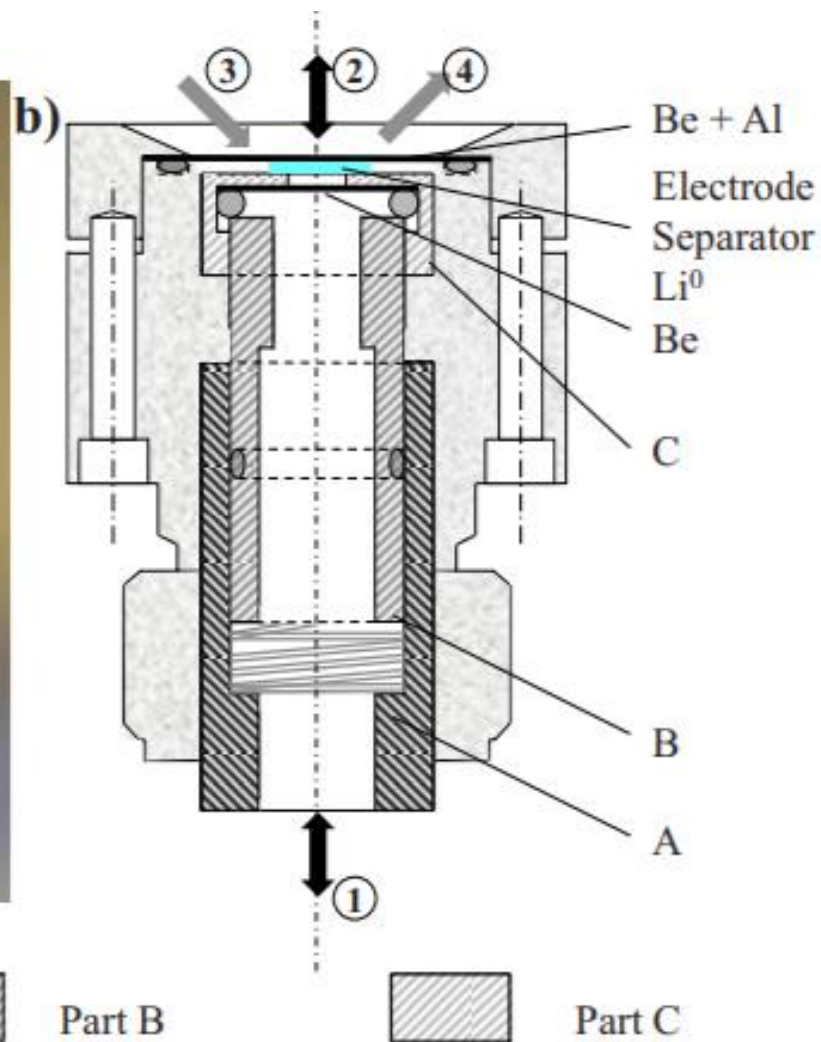
Symmetry most probably changes



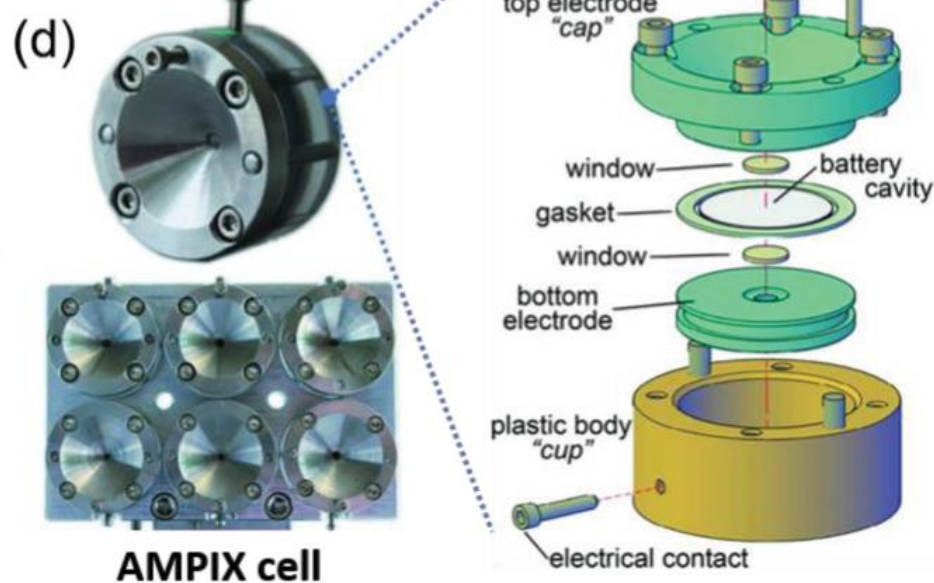
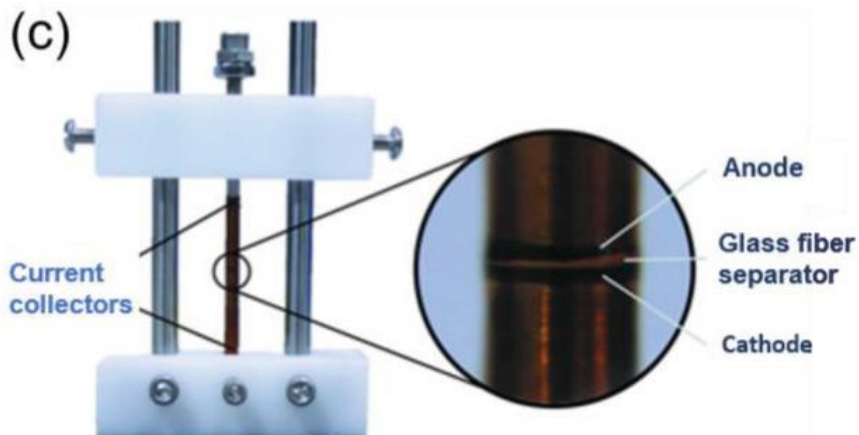
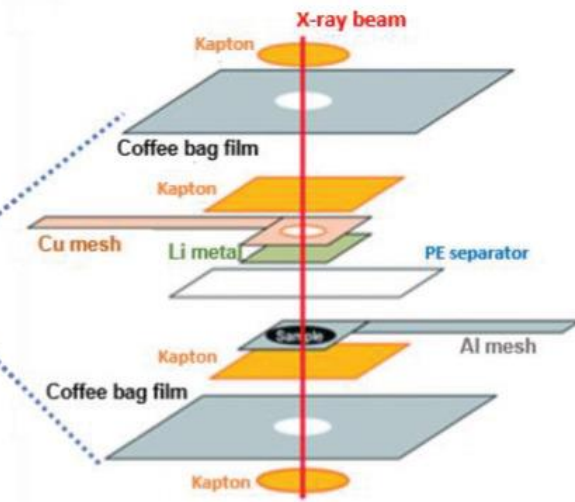
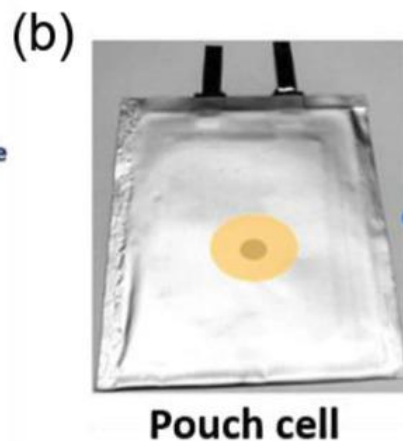
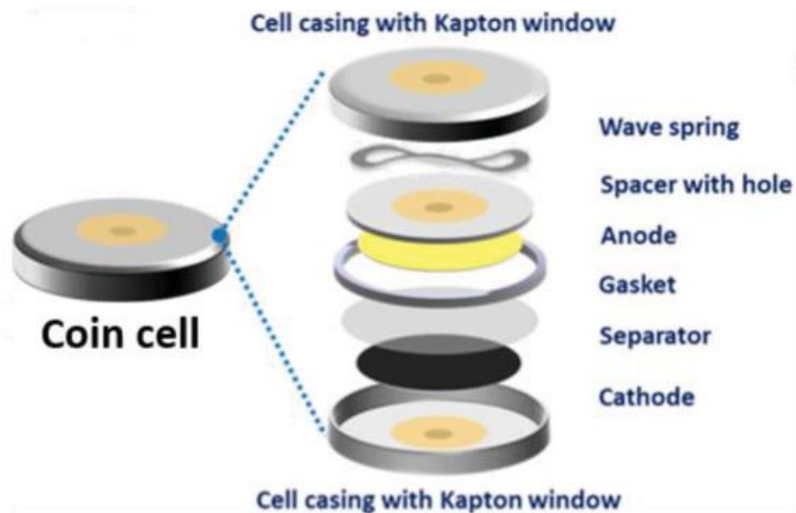
Principle scheme



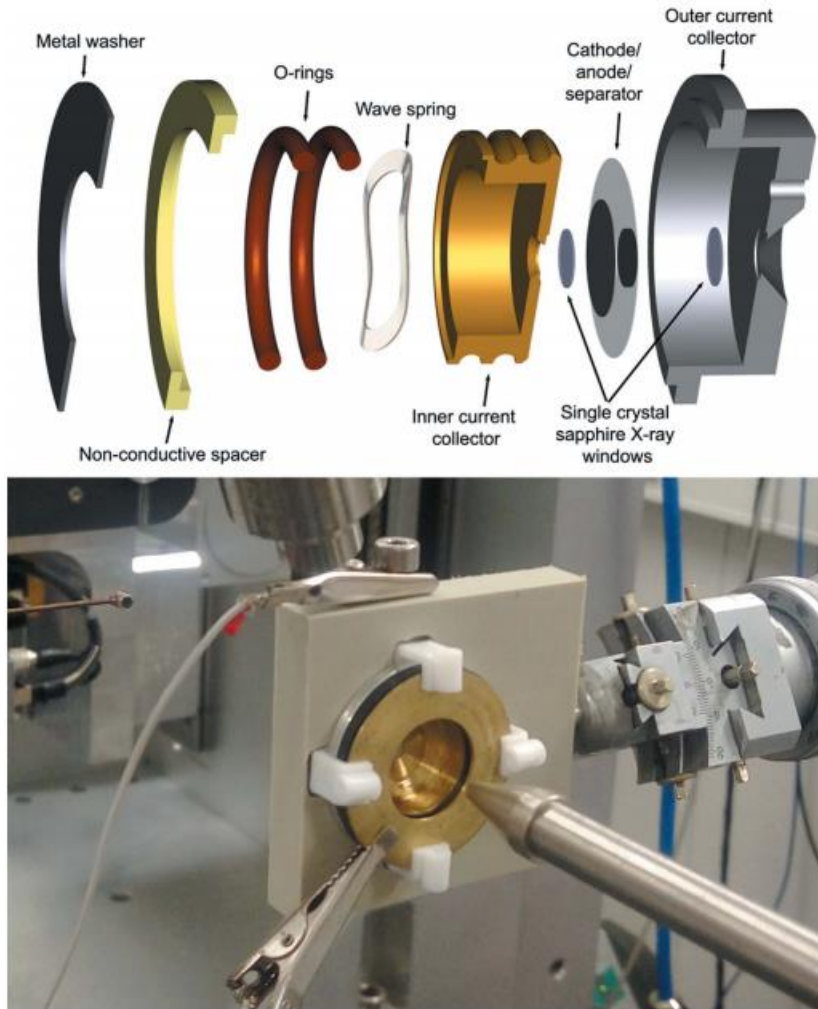
Electrochemical cell



Operando electrochemical cells



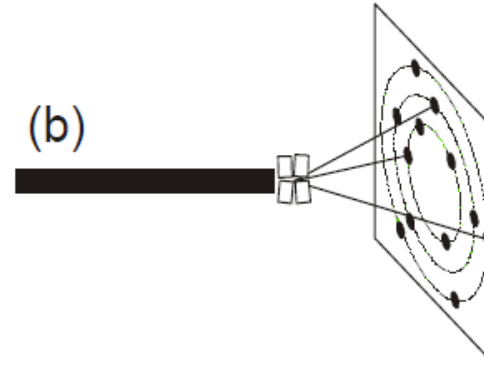
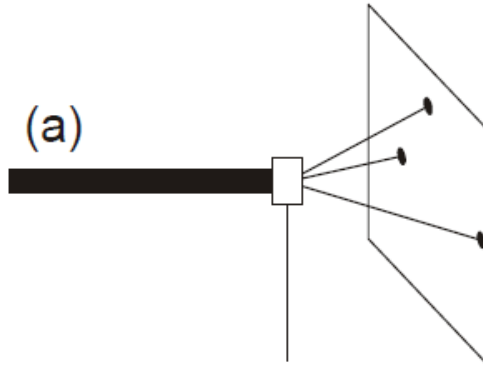
Synchrotron operando cells: ESRF BM01



Signal formation

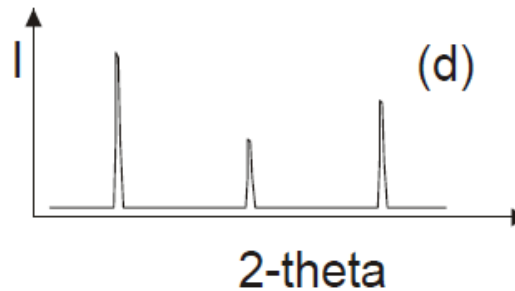
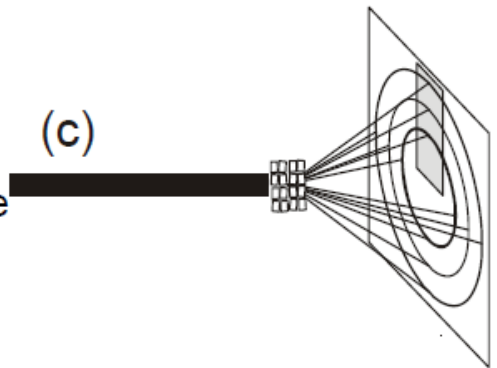


Single crystal

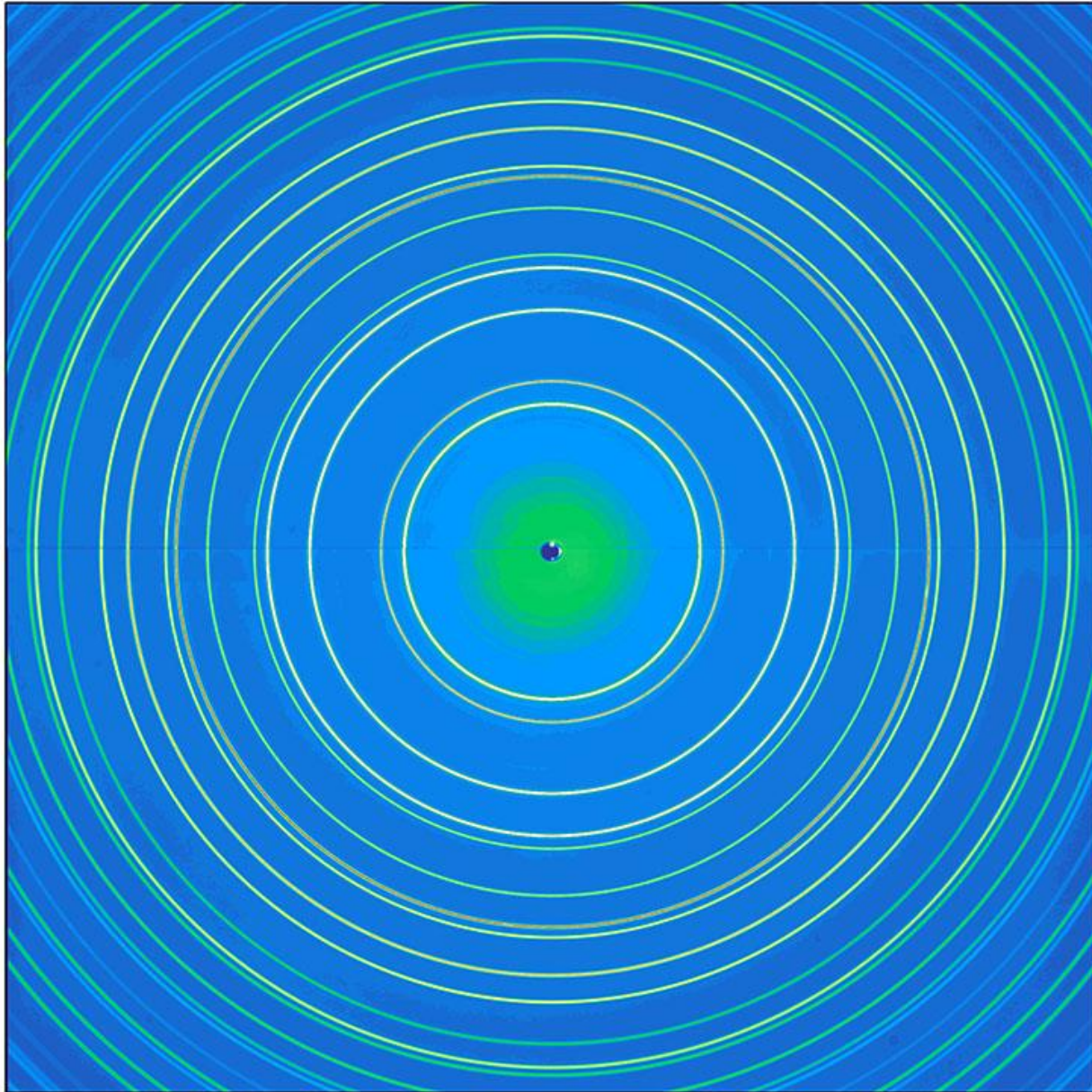


Four differently oriented single crystals

Polycrystalline material

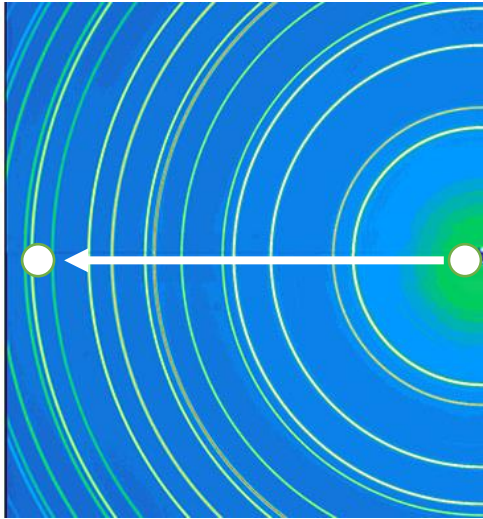


Signal formation



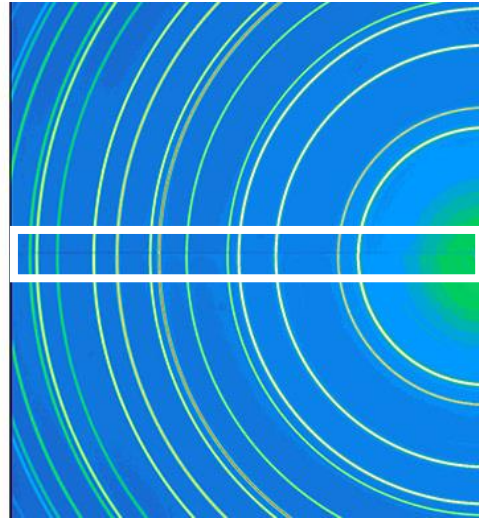
Detector types

0D (scintillation)



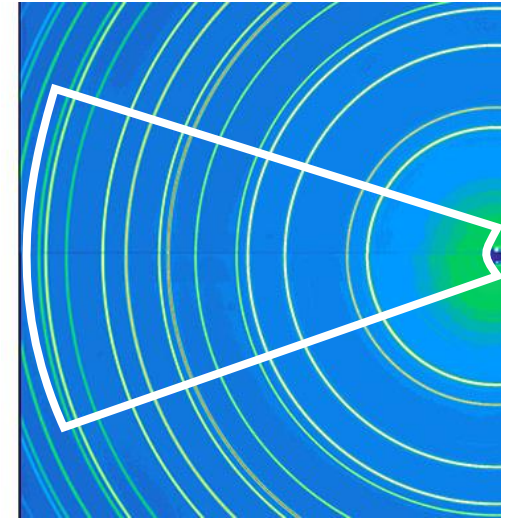
Small spot measured
Scan necessary
Long measuring time

1D (position sensitive, PSD)



Simultaneous
measurement
Medium measuring time

2D



Monocrystals
Oriented samples
Fast measuring time
Instant measuring time

Detector types



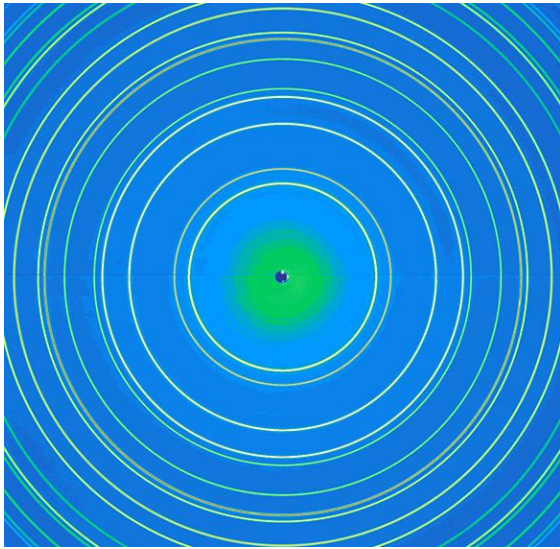
Perkin Elmer



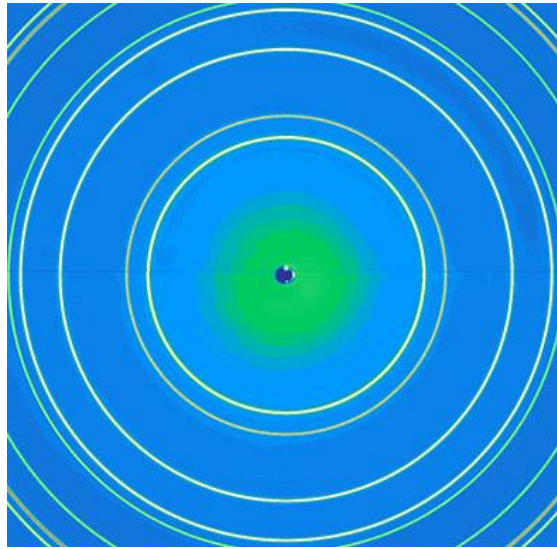
DECTRIS



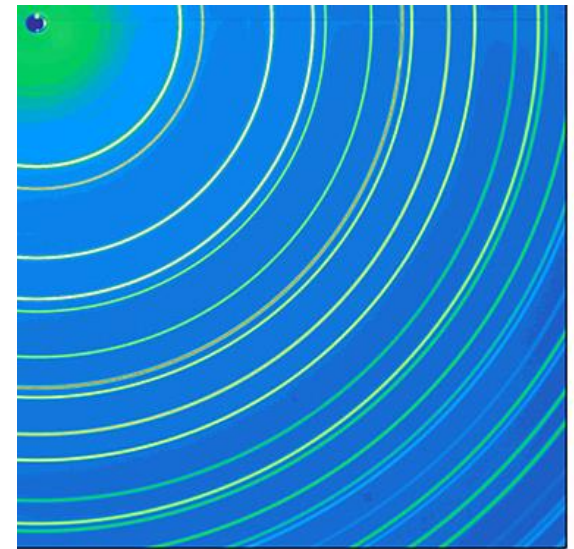
Detector positioning



Close to the sample



Far from the sample



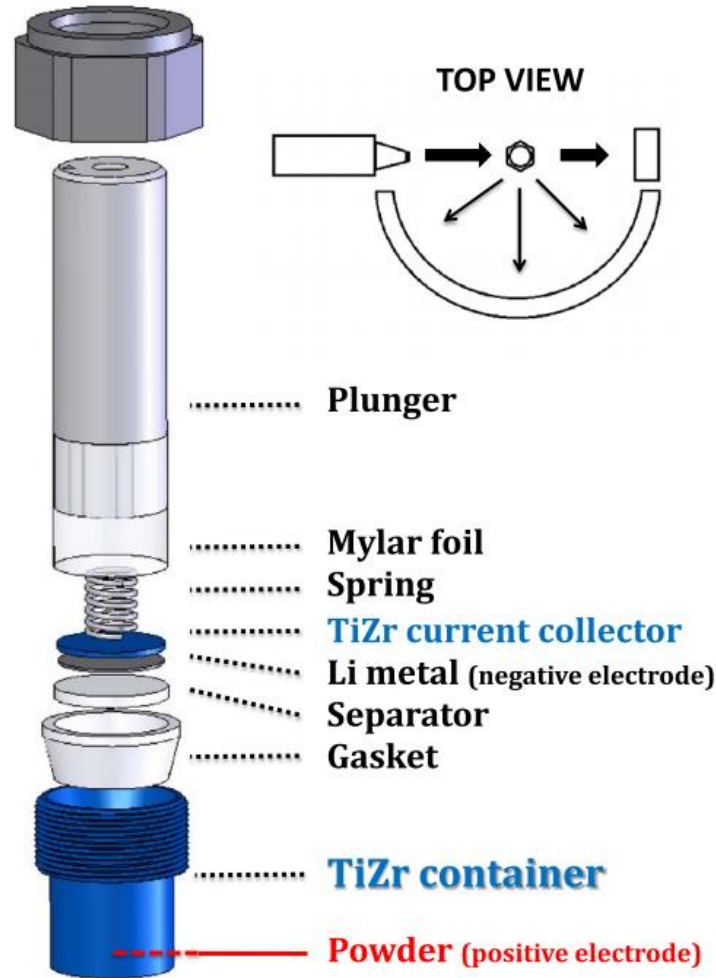
Shifted against the beam

To change the d (Q) space the detector-to-sample distance and detector alignment may be adjusted

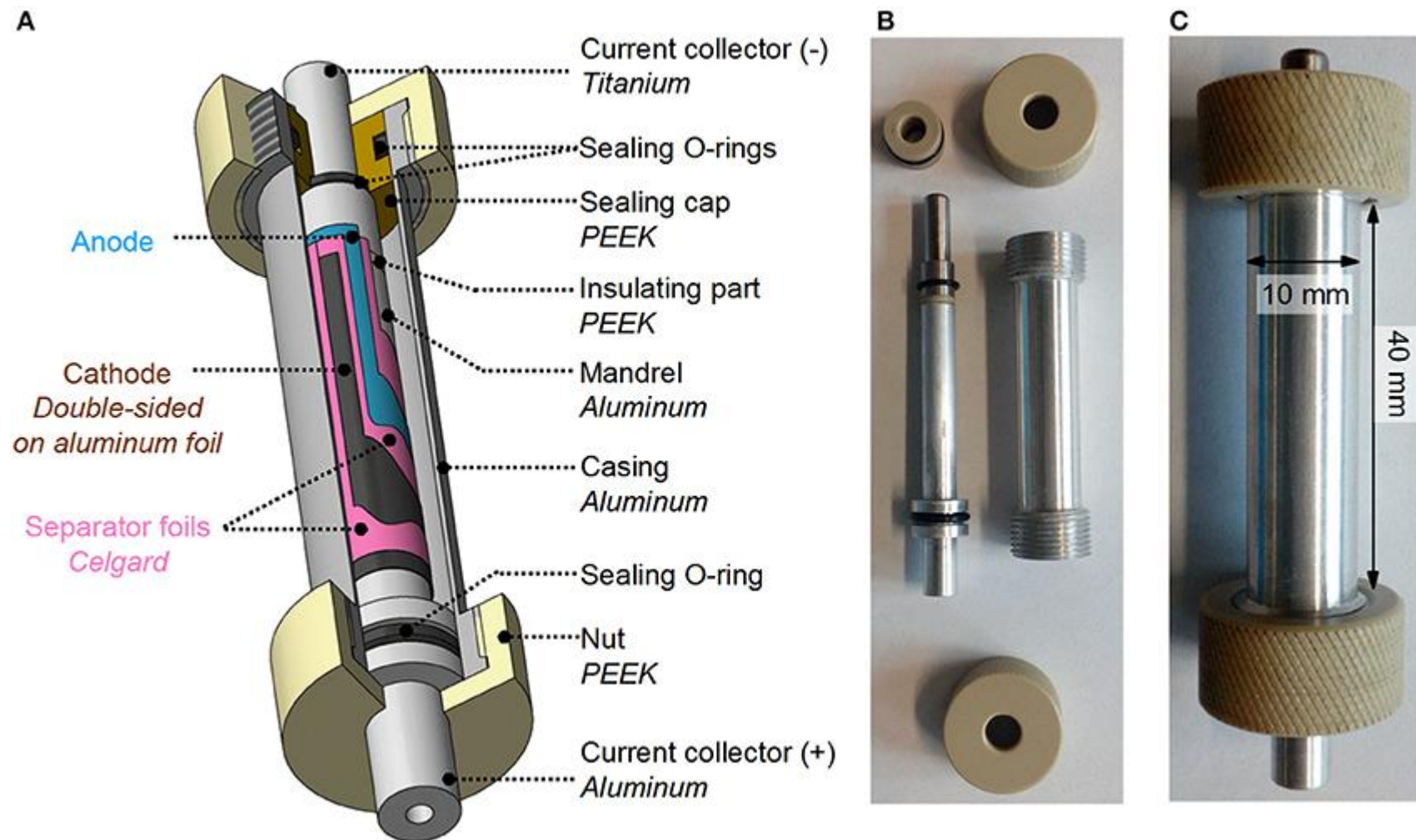
The detector-to-sample distance changes affects the resolution

The detector size matters

Neutron diffraction: operando cells

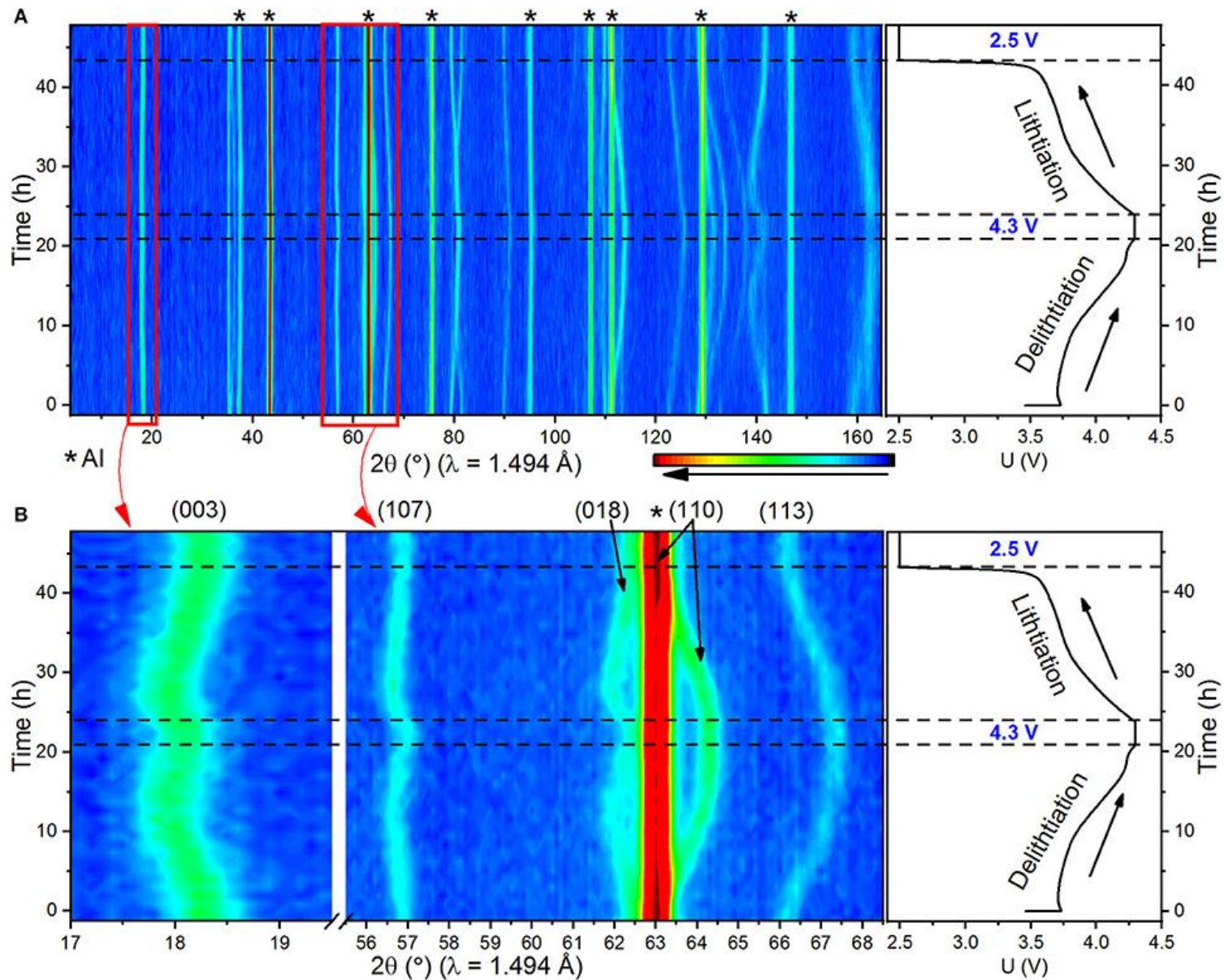


Neutron diffraction: operando cells

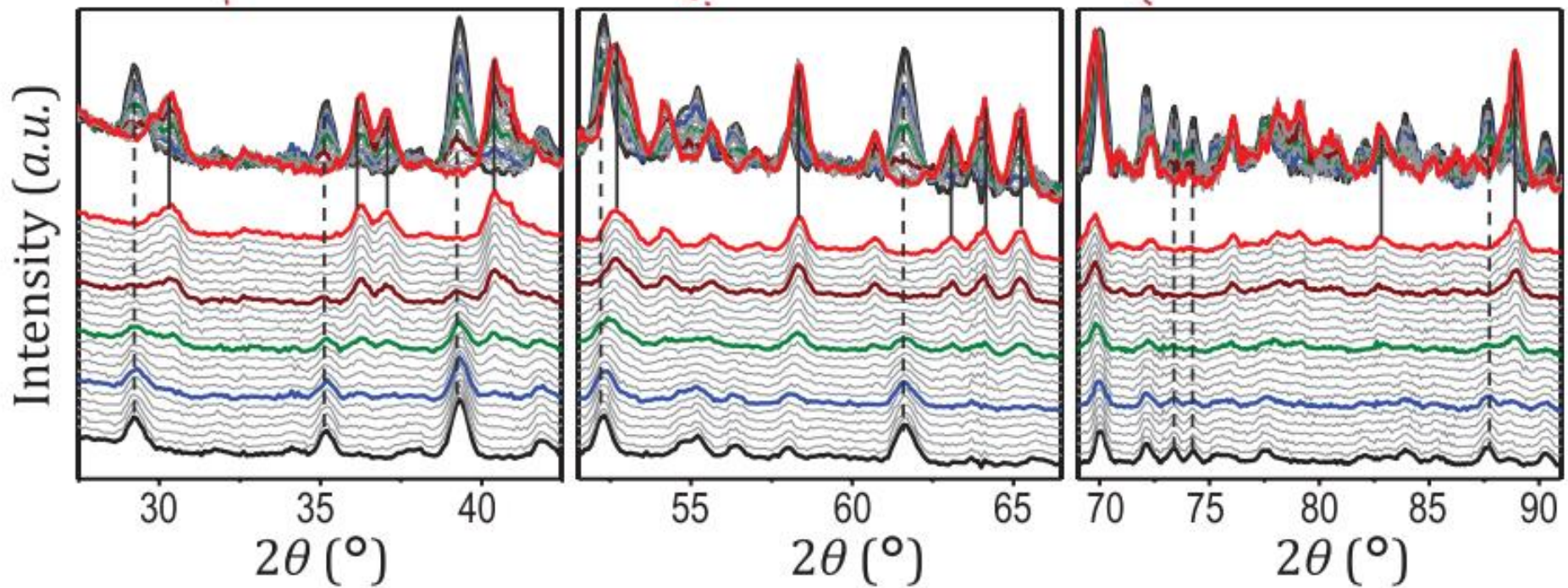


Neutron diffraction NMC 622

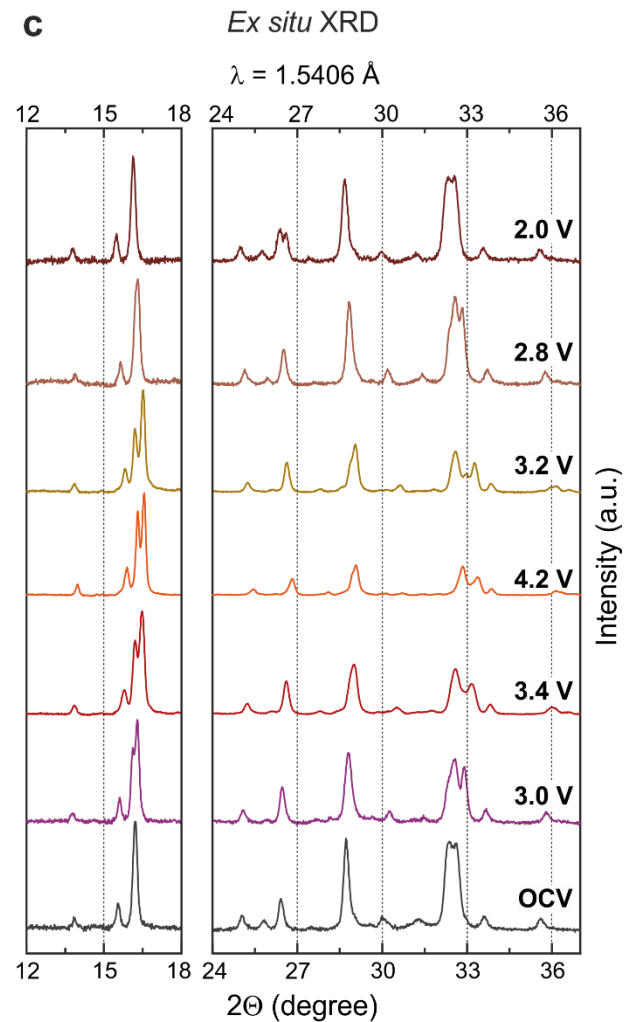
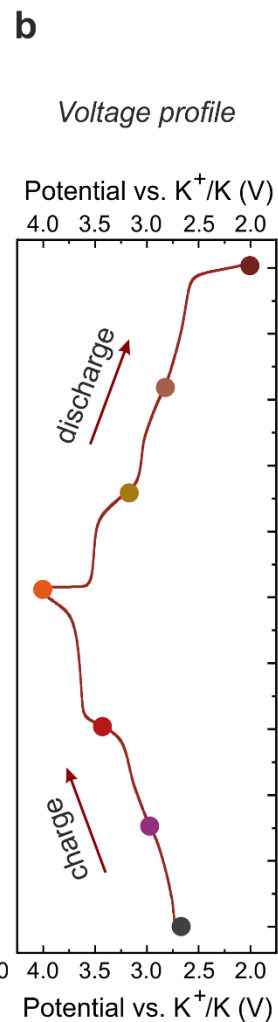
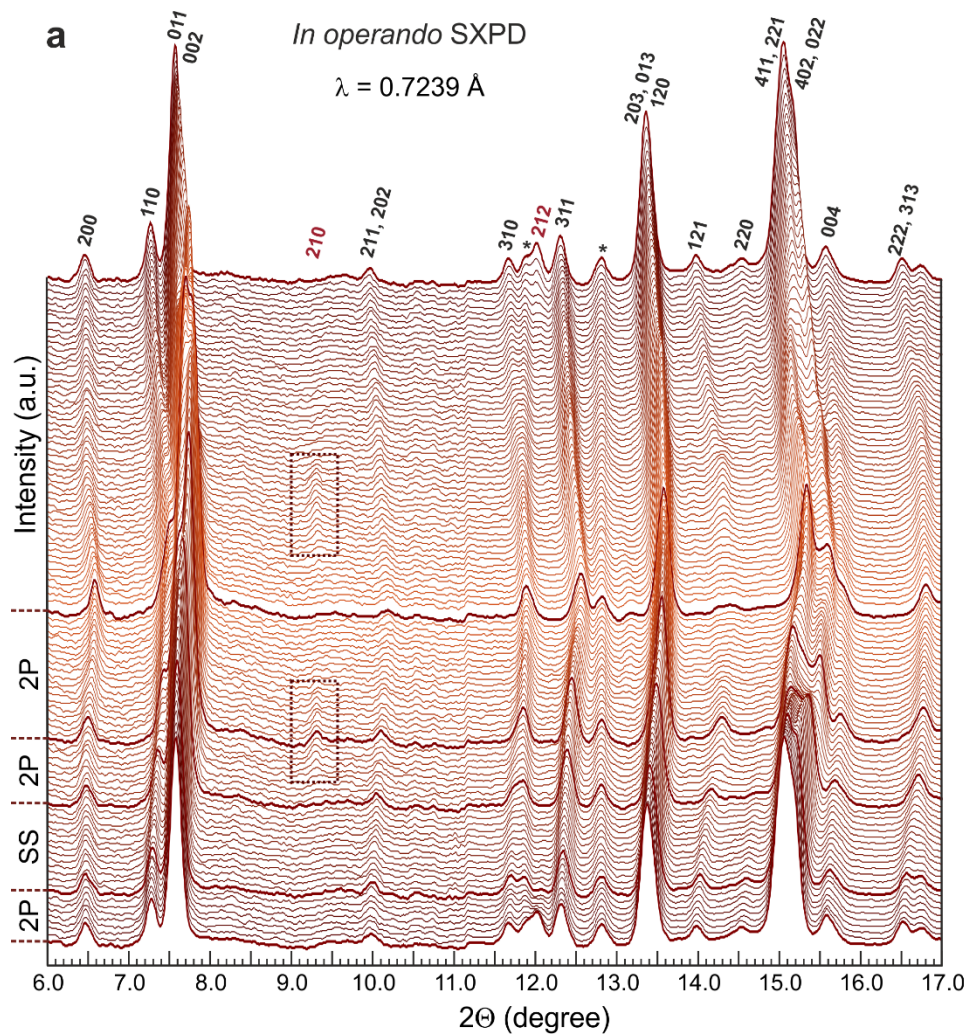
Li(NiCoMn)O₂ 622



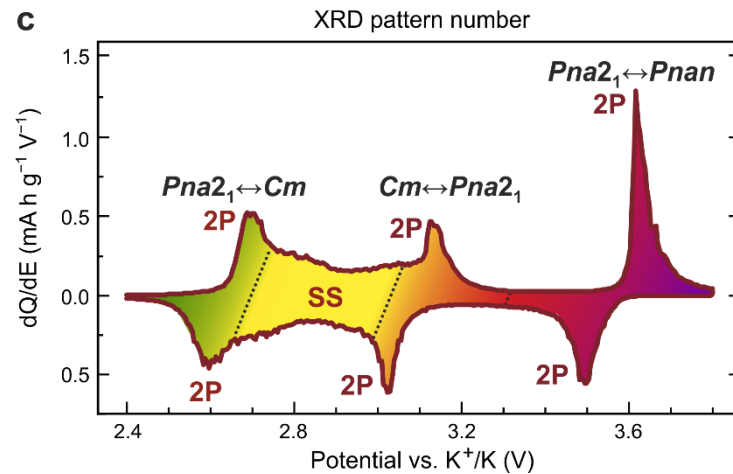
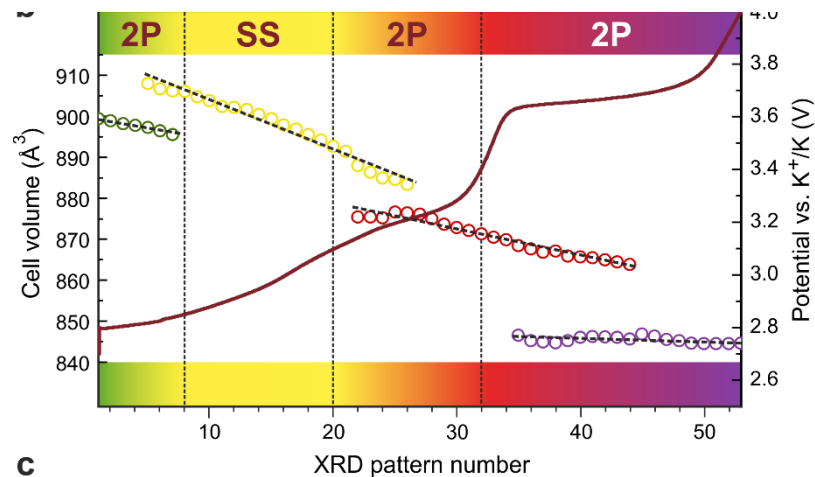
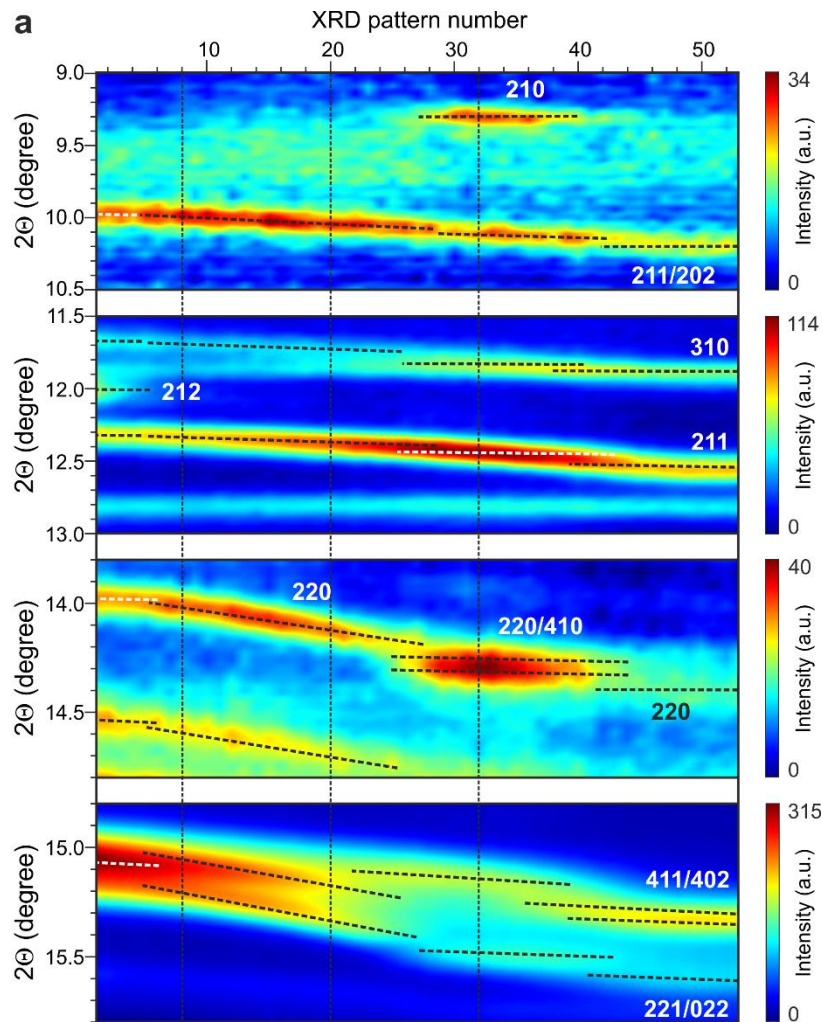
Neutron diffraction



KTiPO₄F



KTiPO₄F



Thank you for your attention!

